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## Key indicators

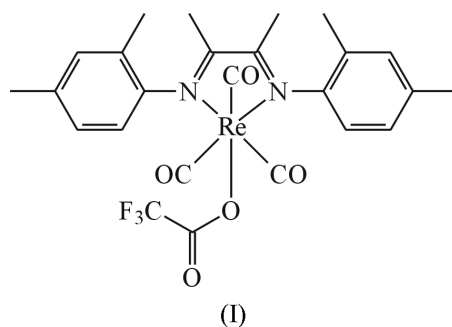
Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.013$  Å  
Disorder in main residue  
 $R$  factor = 0.042  
 $wR$  factor = 0.087  
Data-to-parameter ratio = 16.0For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.**[*N,N'*-Bis(2,4-dimethylbenzene)-2,3-dimethyl-1,4-diazabuta-1,3-diene]-*fac*-tricarbonyl-(trifluoroacetato)rhenium(I)**

In the title compound,  $[\text{Re}(\text{CF}_3\text{COO})(\text{C}_{20}\text{H}_{24}\text{N}_2)(\text{CO})_3]$ , the Re atom has a distorted octahedral configuration. The 1,4-diazabutadiene part of the donor ligand and two carbonyl groups occupy the equatorial plane of the complex, with the third carbonyl ligand and the trifluoroacetate group in the axial positions. The equatorial diimine ligand is bound to the Re center in an *N,N'*-bidentate chelating fashion, leading to a five-membered  $\text{ReN}_2\text{C}_2$  chelate ring with an  $\text{N}-\text{Re}-\text{N}$  bite angle of  $73.4(2)^\circ$ .

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## Comment

We report here the results of an X-ray structure investigation of the title complex, (I). The Re atom has a distorted octahedral configuration (Fig. 1) and is coordinated by two carbonyl groups [ $\text{Re}-\text{C} = 1.900(9)$  and  $1.901(9)$  Å] and the diimine group of the organic ligand [ $\text{Re}-\text{N} = 2.156(5)$  and  $1.157(5)$  Å] in the equatorial plane. Octahedral coordination is completed at the axial positions by another carbonyl ligand [ $\text{Re}-\text{C} = 1.912(8)$  Å] and a carboxyl O atom of the trifluoroacetate anion [ $\text{Re}-\text{O} = 2.157(4)$  Å]. The orientation of the coordinated trifluoroacetate anion can be defined by the  $\text{O}2-\text{C}8-\text{C}9-\text{F}1$  torsion angle of  $-16(2)^\circ$ . The equatorial arrangement, defined by the Re and donor atoms (N1/N4/C5/C6) of the corresponding diimine and carbonyl ligands is essentially planar; the deviation of the Re atom from the N1/N4/C5/C6 plane is  $0.068(3)$  Å.



The equatorial diimine ligand is bound to the Re center in a *N,N'*-bidentate chelating fashion, leading to a five-membered  $\text{ReN}_2\text{C}_2$  chelate ring with an  $\text{N}1-\text{Re}1-\text{N}4$  bite angle of  $73.4(2)^\circ$ . The other *cis*-equatorial angles at Re1 are  $\text{C}5-\text{Re}1-\text{C}6$  of  $88.3(3)^\circ$ ,  $\text{N}1-\text{Re}1-\text{C}5$  of  $98.9(3)^\circ$  and  $\text{N}4-\text{Re}1-\text{C}6$  of  $97.0(2)^\circ$ . The axial  $\text{C}7-\text{Re}1-\text{O}1$  angle is  $177.4(3)^\circ$ . The aromatic rings are oriented essentially at right angles to the chelate ring plane [dihedral angle between  $\text{C}11-\text{C}16$  and  $\text{Re}1/\text{N}1/\text{C}2$  is  $84.9(2)^\circ$  and between  $\text{C}21-\text{C}26$  and  $\text{Re}1/\text{N}4/\text{C}3$  is  $87.4(3)^\circ$ ]; this orientation leads to  $\text{C}121 \cdots \text{O}7$

and C221...O7 intramolecular separations of 3.737 (11) and 3.738 (18) Å, respectively. The CF<sub>3</sub> group has high thermal motion consistent with some unresolved rotational disorder, as is often found in other trifluoroacetate compounds (Rodrigues *et al.*, 2000, 2001).

## Experimental

Compound (I) was synthesized as described in the literature (Klein *et al.*, 1996). Dark red block-shaped crystals of (I) suitable for X-ray structure analysis were obtained by slow evaporation of a solution in dichloromethane/*n*-hexane at room temperature.

### Crystal data

[Re(C <sub>2</sub> F <sub>3</sub> O <sub>2</sub> )(C <sub>20</sub> H <sub>24</sub> N <sub>2</sub> )(CO) <sub>3</sub> ]	$D_x = 1.690 \text{ Mg m}^{-3}$
$M_r = 675.66$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 25 reflections
$a = 12.818 (3) \text{ \AA}$	$\theta = 14.0\text{--}15.0^\circ$
$b = 13.026 (3) \text{ \AA}$	$\mu = 4.63 \text{ mm}^{-1}$
$c = 16.554 (4) \text{ \AA}$	$T = 293 (2) \text{ K}$
$\beta = 106.07 (2)^\circ$	Block, dark red
$V = 2655.9 (11) \text{ \AA}^3$	$0.55 \times 0.30 \times 0.30 \text{ mm}$
$Z = 4$	

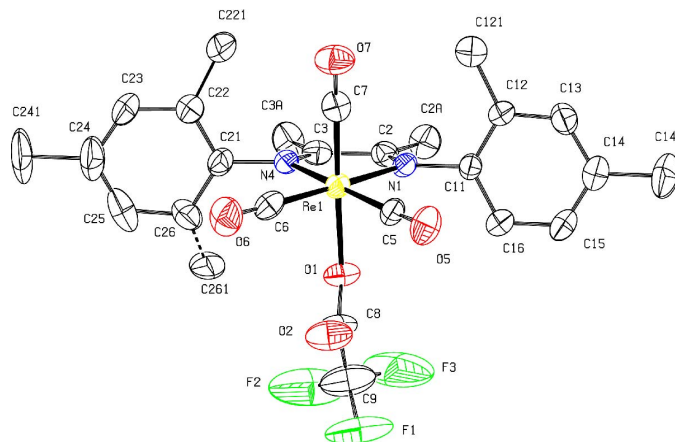
### Data collection

Enraf–Nonius CAD-4 diffractometer	$R_{\text{int}} = 0.015$
$\omega$ -2 $\theta$ scans	$\theta_{\text{max}} = 26.3^\circ$
Absorption correction: $\psi$ scan (DATCOR; Reibenspies, 1989)	$h = 0 \rightarrow 15$
$T_{\text{min}} = 0.137$ , $T_{\text{max}} = 0.250$	$k = 0 \rightarrow 16$
5775 measured reflections	$l = -20 \rightarrow 19$
5369 independent reflections	3 standard reflections
2977 reflections with $I > 2\sigma(I)$	frequency: 60 min
	intensity decay: 3%

### Refinement

Refinement on $F^2$	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.042$	$w = 1/[\sigma^2(F_o^2) + (0.0422P)^2]$
$wR(F^2) = 0.087$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 0.92$	$(\Delta/\sigma)_{\text{max}} = 0.004$
5369 reflections	$\Delta\rho_{\text{max}} = 0.76 \text{ e \AA}^{-3}$
335 parameters	$\Delta\rho_{\text{min}} = -0.63 \text{ e \AA}^{-3}$

One of the *m*-xylyl rings of the diimine ligand has orientational disorder and the refined site-occupancy factors of the disordered parts (methyl groups C221/C226 and associated H atoms) are 0.651 (15) and 0.349 (15). None of the the H atoms of the methyl groups was clearly resolved in difference maps and they were



**Figure 1**

The molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level. Atom C261 is the minor-occupancy methyl C atom. For clarity, H atoms have been omitted.

subsequently included as six half-occupancy H atoms on each group. Other H atoms were located in difference maps and were subsequently repositioned geometrically and refined as riding, with C–H distances of 0.93 (aromatic) and 0.96 Å (methyl), and with  $U_{\text{iso}}(\text{H}) = 1.2$  (aromatic) or 1.5 (methyl) times  $U_{\text{eq}}(\text{C})$ .

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1992); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms, 1996); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97*.

## References

- Enraf–Nonius (1992). *CAD-4 EXPRESS*. Enraf–Nonius, Delft, The Netherlands.
- Harms, K. (1996). *XCAD4*. University of Marburg, Germany.
- Klein, A., Conny, V. & Kaim, W. (1996). *Organometallics*, **15**, 236–244.
- Reibenspies, J. (1989). *DATCOR*. Texas A&M University, USA.
- Rodrigues, V. H., Paixão, J. A., Costa, M. M. R. R. & Matos Beja, A. (2000). *Acta Cryst.* **C56**, 1053–1055.
- Rodrigues, V. H., Paixão, J. A., Costa, M. M. R. R. & Matos Beja, A. (2001). *Acta Cryst.* **C57**, 417–420.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.